

## 4-Nitroaniline–2,4,6-trimethoxybenz-aldehyde (1/1)

Abdullah M. Asiri,<sup>a</sup> Salman A. Khan,<sup>a</sup> Kong Wai Tan<sup>b</sup> and Seik Weng Ng<sup>b\*</sup>

<sup>a</sup>Chemistry Department, Faculty of Science, King Abdul Aziz University, PO Box 80203, Jeddah 21589, Saudi Arabia, and <sup>b</sup>Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia

Correspondence e-mail: seikweng@um.edu.my

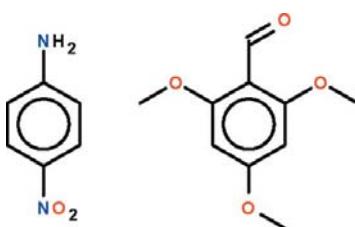
Received 15 June 2010; accepted 18 June 2010

Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(C-C) = 0.003$  Å;  $R$  factor = 0.049;  $wR$  factor = 0.124; data-to-parameter ratio = 12.3.

In the title co-crystal,  $C_6H_6N_2O_2 \cdot C_{10}H_{12}O_4$ , the two components are held together by an  $N-H \cdots O_{\text{aldehyde}}$  hydrogen bond. Adjacent co-crystals are linked by weaker  $N-H \cdots O_{\text{nitro}}$  hydrogen bonds, forming a linear chain. The two aromatic rings of the components are aligned at  $75.2(1)^\circ$ . The crystal studied was a non-merohedral twin with a 24% minor component.

### Related literature

For some examples of co-crystals of 4-nitroaniline, see: Bertolaso *et al.* (2001); Dederer & Gieren (1979); Huang *et al.* (1996); Koshima *et al.* (1996); Rashid & Deschamps (2006); Singh *et al.* (2003); Smith *et al.* (1997); Weber (1981); Zaitu *et al.* (1995). For the treatment of non-merohedral twins, see: Spek (2009).



### Experimental

#### Crystal data

$C_6H_6N_2O_2 \cdot C_{10}H_{12}O_4$

$M_r = 334.32$

Monoclinic,  $P2_1/c$   
 $a = 7.4409(11)$  Å  
 $b = 30.022(5)$  Å  
 $c = 6.9400(11)$  Å  
 $\beta = 93.237(3)^\circ$   
 $V = 1547.9(4)$  Å<sup>3</sup>

$Z = 4$   
Mo  $K\alpha$  radiation  
 $\mu = 0.11$  mm<sup>-1</sup>  
 $T = 100$  K  
 $0.15 \times 0.10 \times 0.05$  mm

#### Data collection

Bruker SMART APEX  
diffractometer  
8127 measured reflections

2722 independent reflections  
1834 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.059$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.049$   
 $wR(F^2) = 0.124$   
 $S = 1.01$   
2722 reflections

221 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.22$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.41$  e Å<sup>-3</sup>

**Table 1**  
Hydrogen-bond geometry (Å, °).

$D-\text{H} \cdots A$	$D-\text{H}$	$\text{H} \cdots A$	$D \cdots A$	$D-\text{H} \cdots A$
N1—H12···O1	0.86	2.16	3.016 (3)	172
N1—H11···O5 <sup>i</sup>	0.86	2.50	3.288 (3)	152
N1—H11···O6 <sup>i</sup>	0.86	2.50	3.293 (3)	154

Symmetry code: (i)  $x + 1, -y + \frac{3}{2}, z + \frac{1}{2}$ .

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2010).

We thank King Abdul Aziz University and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SI2271).

### References

- Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
- Bertolaso, V., Gilli, P., Ferretti, V. & Gilli, G. (2001). *New J. Chem.* **25**, 408–415.
- Bruker (2009). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA..
- Dederer, B. & Gieren, A. (1979). *Naturwissenschaften*, **66**, 470–471.
- Huang, K.-S., Britton, D. & Etter, M. C. (1996). *Acta Cryst. C52*, 2868–2871.
- Koshima, H., Wang, Y., Matsuura, T., Mizutani, H., Isako, H., Miyahara, I. & Hirostu, K. (1996). *Mol. Cryst. Liq. Cryst.* **279**, 265–274.
- Rashid, A. N. & Deschamps, J. R. (2006). *J. Mol. Struct.* **787**, 216–219.
- Sheldrick, G. M. (2008). *Acta Cryst. A64*, 112–122.
- Singh, N. B., Pathak, A. & Frohlich, R. (2003). *Aust. J. Chem.* **56**, 329–333.
- Smith, G., Lynch, D. E., Byriel, K. A. & Kennard, C. H. L. (1997). *J. Chem. Crystallogr.* **27**, 307–317.
- Spek, A. L. (2009). *Acta Cryst. D65*, 148–155.
- Weber, G. (1981). *Z. Naturforsch. B36*, 896–897.
- Westrip, S. P. (2010). *J. Appl. Cryst.* Submitted.
- Zaitu, S., Miwa, Y. & Taga, T. (1995). *Acta Cryst. C51*, 2390–2392.